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ABSTRACT
Yeasts have been used for the heterologous production of a range of enzymes. However, α-L-rhamnosidase production in yeasts as well as its vast potential for biotechnological processes is less reported. α-L-Rhamnosidase is one of the important biotechnologically attractive enzymes in several industrial and biotechnological processes. In food and agriculture industries, the enzyme catalyzes the hydrolysis of hesperidin to release L-rhamnose and hesperidin glucoside, industrial removal of bitterness from citrus juices caused by naringin, and enhancing aroma in grape juices and derived beverages. In pharmaceutical and chemical industries, this enzyme is used in the structural determination of polysaccharides, glycosides and glycolipids, metabolism of gellan, conversion of chloropolysporin B to chloropolysporin C, and production of prunin. Rhamnosidases are extensively distributed in fungi and bacteria while their production from yeast sources is less reported. Yeast rhamnosidase is very important as it is produced in short-duration fermentation, with enhanced shelf life, high thermal stability, capable of retaining juice flavor, and is non-toxic for human consumption. In this review, an attempt has been made to fill up this gap by focusing on production, purification, characterization, structural and molecular biological studies of yeast rhamnosidase and its potential biotechnological applications.

Keywords: Industrial applications, Naringin, Rhamnosidase, Yeast
INTRODUCTION
Yeasts are groups of unicellular fungi that belong to the phylum Dikaryomycota. Two main groups of yeasts classified are Ascomycetes or Basidiomycetes (Barnett et al. 2000). The yeasts are characterized by single cells that reproduce by budding from a narrow or broad base (e.g. *Saccharomyces*) or fission from a broad base (e.g. *Schizosaccharomyces*). Additionally, pseudohyphae or true hyphae or both may be present (Kurtzman and Fell 1998). The application of yeasts is not only restricted to the traditional processes of making bread, wine, and beer, though yeasts are also a rich source of a variety of industrially essential enzymes such as rhamnosidase, amylase, protease, invertase, etc. In this chapter, we targeted the enzyme α-L-rhamnosidases produced by yeast. In addition, recent research has generally focused on yeasts as biocontrol on plant disease management for managing postharvest diseases, mostly of fruit in agriculture sectors (Pimenta et al. 2009).

α-L-Rhamnosidase (3.2.1.40) are exo-type enzymes that remove terminal α-L-rhamnosyl groups at the ends of polysaccharides and glycosides containing L-rhamnose. The enzyme converts the bitter glycoside naringin to the less-bitter prunin by cleavage of an α-(1→2) bond between L-rhamnose and glucose (Magario et al. 2009) (Fig. 1). Many other natural glycosides, which include rutin, quercitrin, hesperidin, diosmin, and terphenyl glycosides, containing terminal α-rhamnose can act as substrates of α-L-Rhamnosidase (Ribeiro 2011). Rhamnosidases are important industrial enzymes of great significance in the current biotechnological area with applications in food (Spagna et al. 2000), agriculture, and pharmaceutical (Monti et al. 2004) industrial processes for the bioconversion of natural or synthetic rhamnosides e.g. the elimination of hesperidin crystals in orange juice, for aroma enhancement of wine, in steroid transformation, in the structural study of bacteria polysaccharide, and in the production of L-rhamnose from glycosides (Gallego et al. 2001; Soria and Ellenrieder 2002; Manzanares et al. 2001; Mutter et al. 1994; Puri and Banerjee 2000).

![Fig. 1 Hydrolysis of naringin into prunin and rhamnose by α-L-rhamnosidase.](image-url)
Several microorganisms use α-L-rhamnosidase to release L-rhamnose as a source of carbon and energy (Twerdochlib et al. 1994). The enzyme has been reported from animal tissues, plants, microbes like yeasts, fungi, and bacteria (Yadav et al. 2010). Despite α-L-rhamnosidases being biotechnologically important enzymes, only a few crude rhamnosidase preparations hesperidinase, and naringinase are commercially available so far. All of these preparations are presently obtained from the genera Aspergillus and Penicillium. In the case of yeast strains, there is a lack of information on the production, purification, characterization, preparation, and evaluation of α-L-rhamnosidase. Only a few yeast strains Saccharomyces, Hansenula, Debaryomyces, Candida, Aureobasidium pullulans (Miklosy and Polos 1995; Rosi et al. 1995; Mcmahon et al. 1999; Yadav et al. 2010), Pichia angusta (Yanai and Sato 2000) and Clavispora lusitaniae (Fig. 2) (Singh et al. 2015 and 2018) have been reported. Yeast α-L-rhamnosidase possesses characteristics of high specificity, increased retention of juice flavor, nutrients, and economic viability with a strong ability to remove the bitter taste from citrus fruits.

There is only one review available on α-L-rhamnosidases by Yadav et al. 2010. Therefore, our responsibilities to research and focus on this area and provide more information on rhamnosidases. In this article, an attempt has been made to review the recent literature and summary of yeast rhamnosidases and their potential applications.

**Sources**

Naringinase (α-L-rhamnosidase and β-D-glucosidase) has been reported in the literature since the earliest of 1938, from celery seeds (Hall1938). Other plant sources of α-L-rhamnosidase are grapefruit leaves (Thomas et al.1958), Rhamnus daurica (Suzuki 1962), and Fagopyrum esculentum (Bourbouze et al. 1976). The α-L-rhamnosidases from only two animal sources, viz. Turbo cornutus liver and pig liver have been reported (Kurosawa et al. 1973; Qian et al. 2005). Microorganisms are the main sources of α-L-rhamnosidase, mainly filamentous fungi such as Aspergillus, Circinella, Eurotium, Fusarium, Penicillium, Rhizopus, and Trichoderma (Scaroni et al. 2002). The induction of α-L-rhamnosidases production in several fungal strains such as Acremonium persicinum, Circinella muscae, Emericella nidulans, Fusarium oxysporum, Mortierella alpina, Penicillium oxalicum, Rhizopus arrhizus, Talaromyces flavus, and Trichoderma harzianum, using L-rhamnose, naringin, rutin, hesperidin as inducers (Monti et al. 2004). Aspergillus niger and Penicillium decumbens are the most commonly used species for their production with a potential value in oenology, and their enzymatic activities have been well characterized (Gallego et al. 1996; Manzanares et al. 2003; Orejas et al. 1999; Singh et al. 2015).
The first bacterial $\alpha$-L-rhamnosidase was purified from the genus Bacteroides (Jang and Kim 1996). Other bacterial strains which produce $\alpha$-L-rhamnosidases are thermophilic bacterium (Birgisson et al. 2004), Fusabacterium (Park et al. 2005), Pseudoalteromonas species, Ralstonia pickettii (Orrillo et al. 2007), Lactobacillus acidophilus (Beekwilder et al. 2009), Pediococcus acidilactici (Michlmayr et al. 2011), Clostridium stercolarium (Zverlov et al. 2000), Sphingomonas paucimobilis (Hashimoto and Murata 1998), Bacillus sp. (Hashimoto et al. 1999) and Corticium rolfsii (Kaji and Ichimi 1973). In a few reports, low levels of activity have been found in yeast belonging to genus Saccharomyces, Hansenula, Debaryomyces, Candida, and Aureobasidium (McMahon et al. 1999; Miklosy and Polos 1995; Rosi et al. 1995; Yadav et al. 2010), P. angusta (Yanai and Sato 2000), Clavispora lusitaniae (KF633446), Clavispora lusitaniae (KF633447), Candida sp. YS12A (KF680225) and Candida hyderabadensis (KF680226) (Singh et al. 2015). All yeast strains producing $\alpha$-L-rhamnosidase are summarized in Table 1.

### Assay methods

There are various methods available for measuring $\alpha$-L-rhamnosidase activity, some are mentioned below:

1. Romero et al.'s (1985) method using p-nitrophenyl $\alpha$-L-rhamnopyranoside as a substrate is a convenient and most common method for assaying $\alpha$-L-rhamnosidase activity. The absorbance of the mixture is determined spectrophotometrically at 400 nm ($A_{400}$). The amount of enzyme that released 1 µmol of p-nitrophenol in 1 min is defined as one unit of $\alpha$-L-rhamnosidase activity ($E400_{nm} p$-nitrophenolate= 21.44 mM$^{-1}$cm$^{-1}$) (Fig. 3). The use of a synthetic substrate did not affect the pH, temperature, or ionic strength optima of the enzyme (Puri 2012).
2. Spectrophotometric determination of flavanones according to the alkaline diethylene glycol method of Davis (1947) is used to evaluate naringinase (α-L-rhamnosidase + β-D-glucosidase) activity. In this method, naringin reacts with diethylene glycol in an alkaline solution to produce a yellow color which is measured at a wavelength of 420 nm ($A_{420}$) (Davis 1947). 2,4-Dinitrosalicylic acid (DNS) method through rhamnose and glucose determination was also applied for naringinase activity (Miller 1959).

3. Ribeiro and Ribeiro (2008) developed an effective HPLC-PAD method for the simultaneous determination of naringin, prunin, and naringenin. The method was linear, precise, and selective for naringin and naringenin identification and quantification (Puri 2012).

4. The enzyme activity toward naringin and hesperidin was assayed for determining rhamnose release by using high-performance anion-exchange chromatography with pulsed amperometric detection (HPAEC-PAD) on a CarboPac PA1 anion-exchange column. One unit of rhamnosidase activity was defined as the amount of enzyme required to release rhamnose at 40 °C and pH of 4.0 at the rate of 1 μmol min$^{-1}$ (Koseki et al. 2008).

5. Assays with nonchromogenic substrates (naringin, hesperidin, rutin, and rutinose) were conducted by Michlmayr et al. (2011). Rhamnose, glucose, and rutinose were quantified by HPLC analysis with a CarboPac PA 1 column and pulsed amperometric gold electrode detection. One unit of rhamnosidase activity is expressed as released rhamnose (μmol) per min at 37°C and pH 5.5.

6. In few reports, L-rhamnose was separated using TLC and quantified to measure α-L-rhamnosidase activity by Victor et al. 1987 and Feng et al. 2005.

**Yeast rhamnosidase**

The fungal and bacterial sources of α-L-rhamnosidases have been thoroughly explored; although yeast rhamnosidase has rarely and is less studied and needs to explore more extensive research studies. At present, there are only one α-L-rhamnosidases of yeast origin *Pichia angusta* X349 that has been purified and characterized (Yanai and Sato 2000). Here we documented some reports on α-L-rhamnosidase activity from yeast that have been published (Table 1).

**Production, purification, and characterization**

Optimization of rhamnosidase production from *Clavispora lusitaniae* KF633446 by using multivariate response surface methodology was studied by Singh et al. (2018). The observed optimized parameter for highest rhamnosidase production in the minimal medium was (w/v) 0.6% rhamnose, 0.4% yeast extract, 35±5 °C temperature, and 4 pH.

Yanai and Sato (2000) examined 386 yeast strains for α-L-rhamnosidase activity using L-rhamnose as the inducer. The screening results showed that α-L-rhamnosidase activity was generally restricted to species *S. cerevisiae*, *C. terreus*, *P. angusta*, and *P. capsulata*. One strain, *P. angusta* X349, was selected as the best producer of the enzyme. Selected strain *P. angusta* X349 has been purified to homogeneity using ammonium sulphate precipitation and column chromatography on concanavalin A-Sepharose, DEAE Bio-Gel A agarose, Rhamnose-Sepharose 6B, and hydroxyapatite. The *P. angusta* X349 rhamnosidase was a monomeric protein with a molecular mass of 90 kDa and the isoelectric point at 4.9. The enzyme was opti-
mally active at pH 6.0 and around 40 °C. The \( K_i \) for L-
rhamnose inhibition was 25 mM. The enzyme was
inhibited by \( \text{Cu}^{2+}, \text{Hg}^{2+}, \) and \( \text{p}-\text{chloromercuribenzoate}. \)
The \( \alpha \)-L-rhamnosidase was highly specific for \( \alpha \)-L-
rhamnopyranoside and liberated rhamnose from nar-
ingin, rutin, hesperidin, and 3-quercitin.

The \( \alpha \)-L-rhamnosidase activity of *Pichia guillier-
mondii* NPCC1053 indigenous wine strain from the
North-Patagonian region was evaluated on various
culture media (Rodriguez et al. 2010). The \( \alpha \)Rh activi-
ty of *P. guilliermondii* was associated with its cellular
growth and the maximum value of total enzymatic ac-
tivity (33 U/mg dry weight) under optimal conditions
(YNB- rhamnose supplemented with ammonium sul-
phate) was at the end of the exponential growth phase
(sixth day). This study contributes to the knowledge of
\( \alpha \)-L-rhamnosidases from yeast origin and the enzyme
that could be used in the production of more aromatic
young wines.

A total of thirty yeast strains were isolated from a
whey beverage and screened for \( \alpha \)-L-rhamnosidase
enzyme production (Singh et al. 2015). Of these, only
four isolates *Clavispora lusitaniae* (KF633446), *Clav-
isporea lusitaniae* (KF633447), *Candida* sp. YS12A
(KF680225) and *Candida hyderabadensis* (KF680226)
were capable of producing the \( \alpha \)-L-rhamnosidase en-
zyme by hydrolyzing naringin. The production of \( \alpha \)-L-
rhamnosidase varied in the range of \( 0.029 \pm 0.00 \) to
\( 0.058 \pm 0.02 \) U mL\(^{-1}\). *Clavispora lusitaniae* KF633446
was the best producer of an enzyme. The enzyme was
most active at pH 4 and temperature 50 °C using p-
nitrophenyl-\( \alpha \)-L-rhamnopyranoside as the substrate
(Singh et al. 2015). Maximum activity at pH 4 is close
to that of fruit juice, making it suitable for use in fruit
juices and wine.

*Pichia pastoris* is presently one of the most favored
microorganisms for recombinant enzyme production
owing to its proficient expression system. The upscal-
ing of recombinant \( \alpha \)-L-rhamnosidase production
cloned in the Mur\(^8\) strain of *P. pastoris* KM71H was
optimized by simplified methanol feeding protocol.
Results showed that the specific activity of \( \alpha \)-L-
rhamnosidase improved from 35 U/mg\(^{-1}\) up to 82 U/mg\(^{-1}\)
in the upscaled fermentation, which is the maximum
specific activity of other recombinant and also wild
producers described so far (Markosova et al. 2015).

**Molecular and structural biology**

Only some reports on the molecular and structural
properties of \( \alpha \)-L-rhamnosidase are available. The ram
A is the first cloned \( \alpha \)-L-rhamnosidase gene obtained
from the anaerobe thermophilic bacterium, which be-
longs to the new type of glycose hydrolase family
(Zverlov et al. 2000). Several microorganisms includ-
ing yeast use \( \alpha \)-L-rhamnosidase to release L-rhamnose
as a source of carbon and energy. In the oxidative
pathway for the catabolism of L-rhamnose, L-
rhamnose is first oxidized to L-rhamnono-\( \gamma \)-lactone by
a NAD-utilising L-rhamnose-1-dehydrogenase. This
inducible enzyme activity has been reported in various
yeast species including *Scheffersomyces stipitis*, *Aure-
obasidium (Pullularia) pullulans*, and *Debaryomyces
polymorphus* (Rigo et al. 1976; Vieira et al. 1979;
Twerdochlib et al. 1994). Koivistoinen et al. (2012)
studied the characterization of the gene cluster for L-
rhamnose catabolism in the yeast *Scheffersomyces
(Pichia) stipitis*. The genes for L-rhamnose catabolism
RHA1, LRA2, LRA3, and LRA4 but not LADH are
clustered in *Scheffersomyces (Pichia) stipitis* and relat-
ed fungal species, located next to the cluster is a tran-
scription factor, TRC1, which is conserved among re-
lated species. Transcriptome analysis shows that all the
catabolic genes and all genes in the cluster are up-
regulated on L-rhamnose. The LRA4 gene is not part
of the cluster and it has several paralogues in L-
rhamnose utilizing yeasts, so the function of one of the
paralogues, LRA41 by heterologous expression and
biochemical characterization was also studied by them.

To date, four structures have been resoluted from the
GH78 rhamnosidase family. The first crystal structure
of \( \alpha \)-L-rhamnosidase RhaB from the *Bacillus* sp. GL1
is accessible at 1.9Å resolution. This protein is ho-
modimeric, consisting of four \( \beta \)-sandwich domains, a
core catalytic (\( \alpha/\alpha \) \( \beta \)) barrel, contains 956 amino acid
residues, and 106 kDa molecular mass (Cui et al.
2007). The second structure of *Streptomyces avermiti-
is* (SaRha78A; PDB code 3W5N) \( \alpha \)-L-rhamnosidase
was determined in complex with L-rhamnose, this pro-
tein is large, monomeric, and consisting of six domains
(Fujimoto et al. 2013). The third structure, a putative \( \alpha \)
-L-rhamnosidase from *Bacteroides thetaiotaomicron* VP1-5482 (BT1001; PDB code 3CIH), determined in a structural genomics project is unpublished, also a homodimer. More recently, the crystal structure of KoRha, a putative α-L-rhamnosidase from *Klebsiella oxytoca* has been determined at 2.7 Å resolution with rhamnose bound in the active site of the catalytic domain. The structure reveals an elongated homodimer which is significantly smaller than those of the other earlier determined GH78 structures. Asp 222, the putative catalytic acid, is preceded by an unusual non-proline cis-peptide bond which helps to project the carboxyl group into the active center (Neill et al. 2015).

No information on the three-dimensional structure of an α-L-rhamnosidase from yeast is available. So, there is a scientific demand to isolate the gene coding for α-L-rhamnosidase and crystallize α-L-rhamnosidases from yeast and other sources. Structure analysis of family GH78 enzymes will help to clarify their mechanisms of catalysis and substrate specificity and to improve their potential application in a wide variety of industries.

**Industrial applications of α-L-rhamnosidase**

α-L-Rhamnosidases have several technical applications in the food, chemical, and pharmaceutical industries for the bioconversion of natural or synthetic rhamnosides (Table 2).

**Table 2.** Biotechnological applications of α-L-rhamnosidase enzyme.

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<th>Applications</th>
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<td><strong>Food and Agri-culture</strong></td>
<td><strong>Beverages</strong>&lt;br&gt;Debittering of fruit juices&lt;br&gt;Removal of hesperidin crystals&lt;br&gt;Aroma enhancement in wine&lt;br&gt;Additives&lt;br&gt;Gellan depolymerization&lt;br&gt;Tomato pulp digestion</td>
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<tr>
<td><strong>Chemical</strong></td>
<td><strong>Naringin extraction</strong>&lt;br&gt;Rhamnose preparation&lt;br&gt;Glycolipid’s production</td>
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<tr>
<td><strong>Pharmaceutical</strong></td>
<td><strong>Lectin-directed enzyme activated prodrug therapy (LEAPT)</strong>&lt;br&gt;<strong>Steroid’s biotransformation</strong>&lt;br&gt;<strong>Ginsenosides production</strong>&lt;br&gt;<strong>Antibiotics preparation</strong>&lt;br&gt;<strong>Prunin preparation</strong>&lt;br&gt;<strong>Flavonoids deglycosylation</strong></td>
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**FOOD AND AGRICULTURE INDUSTRY**

**Beverages**

**Debittering of fruit juices**

Citrus juice turns bitter after extraction due to the chemical naringin (flavanoid) and limonin (limonoid). Naringin is the major component in citrus fruit with a very bitter taste. The enzyme naringinase is composed of α-L-rhamnosidase and β-D-glucosidase. Naringin (4,5,7'-trihydroxyflavonone-7′-rhamnoglucoside) is first hydrolyzed by an α-L-rhamnosidase activity of naringinase to rhamnose and prunin (one-third of the bitterness of naringin) which can be further hydrolyzed into glucose and naringenin by the β-D-glucosidase component of naringinase (Singh et al. 2015). Rhamnose and prunin molecules have great potential, especially in the food and pharmaceutical industries (Fig. 1). Singh et al. (2015) produced debittered kimno beverage using α-L-rhamnosidase producing *Clavispora lusitaniae* KF633446 yeast. The decrease of naringin with storage was 443.00 ± 10.00 to 143.70 ± 4.00 ppm due to the α-L-rhamnosidase activity.

**Removal of hesperidin crystals**

The crystallization of soluble hesperidin in the canned mandarin orange juice causes turbidity to the juice (Baker and Tatum 1986). The hesperidinase enzyme-containing α-L-rhamnosidase activity produced by several fungi, especially *Aspergillus niger* is used to prevent the presence of hesperidin crystals in citrus products (Soares and Hotchkiss1998). Hesperidin and hesperidin glycosides treated with α-L-rhamnosidase are highly soluble in water and even stored for a long period without crystal precipitation (Miyake and Yumatomo 1999).

**Aroma enhancement in wine**

Rhamnosidase (with β-glucosidase and arabinosidase) is used for aroma enhancement in winemaking. Monoterpenes such as geraniol, linalool, and α-terpineol are the most important flavoring substances in alcoholic beverages such as wine (Marais 1983; Mateo and Jiménez 2000) and sweet potato shochu (Ohta et al. 1990). A large number of mono-terpenes are present in odorless diglycoside conjugates, the hydrolysis of these diglycosides occur in two enzymatic steps (Günata et al. 1988; Zverlov et al. 2000). In a first reaction, depending on the conjugate, the glyco-
sidic linkage is cleaved by either β-D-apisidase, α-L-arabinofuranosidase, or an α-L-rhamnosidase releasing the β-D-glucoside. In a second reaction, the β-D-glucoside is hydrolyzed by the action of a β-D-glucosidase causing the release of glucose and a volatile compound aromatically active (Gunata et al. 1988; Gunata 2002; Rodríguez et al. 2010). α-L-Rhamnosidase from P. guilliermondii NPCC1053 is an indigenous yeast strain that was able to release monoterpenols and alcohols from grape glycosidic extracts (Rodriguez et al. 2010). P. angusta α-L-rhamnosidase may be useful in winemaking due to the high tolerance to glucose and ethanol, their specificity for the aglycone moieties of grape glycosides, and their hydrolytic activities to natural flavonoid rhamnoglycosides (Yanai and Sato 2000).

Additives
Gellan depolymerization
Gellan, a bacterial exopolysaccharide, has broad application in the food industry owing to its high viscosity in the presence of divalent cations (Cui et al. 2006). Used in the preparation of food additives from biopolymers as well as in the preparation of sweeteners (Giavasis et al. 2000). Bacillus sp. GL1 was isolated to enzymatically modify the viscous properties of gellan gum (Hashimoto et al. 1999) and has a set of enzymes including an α-L-rhamnosidase responsible for the complete depolymerization of gellan gum (Cui et al. 2006).

Tomato pulp digestion
Bacterial α-L-rhamnosidase activity was first attributed to the gut bacteria, which together with β-glucosidase convert ingested flavonoid glycosides into their aglycone forms (Griffiths and Barrow 1972, MacDonald et al. 1983). The rhamnosidases from L. planta-

Chemical industry
Naringin extraction
In industrial citrus processing, solid waste products are obtained, they are essentially constituted of peel and pulp. The citrus solid waste products are rich in fermentable materials as sugars and pectins, high concentrations of flavanone glycosides, hesperidin, and naringin, and lower amounts of many other flavonoids. The purified α-L-rhamnosidase established hydrolysis of naringin extracted from kinnow peel thus endorses its industrial applicability for producing rhamnose (Puri et al. 2011).

Rhamnose preparation
α-L-Rhamnosidase had potential in the manufacture of L-rhamnose by hydrolysis of natural glycosides containing terminal L-rhamnose (Cheetham and Quail 1991) (Fig. 1). Recombinant α-L-rhamnosidase has industrial applicability for the production of rhamnose and prunin from citrus peel waste (Kaur et al. 2010).

Glycolipids production
Glucolipid production from Candida bombicola sophorolipids by P. decumbens naringinase (α-L-rhamnosidase + β-D-glucosidase) showed that the enzyme might be useful for the production of special fatty acids (Saerens et al. 2009).

Pharmaceutical Industry
Lectin-directed enzyme activated prodrug therapy
The lectin-directed enzyme-activated prodrug therapy (LEAPT) bipartite drug delivery system utilizes glycosylated enzyme, localized according to its sugar pattern, and capped prodrugs released by that enzyme. Prodrugs of doxorubicin and 5-fluorouracil capped by the nonmammalian L-rhamnosyl sugar unit have been efficiently synthesized and evaluated for use in the LEAPT system, released by synthetically rhamnosidase enzyme (Robinson et al. 2004; Garnier et al. 2010).

Steroids biotransformation
The hydrolysis of diosgene (a saponin) by α-L-rhamnosidase produces α-L-rhamnose and diosgenin which is used in the synthesis of clinically useful steroid drugs such as progesterone (Elujoba and Hardman 1987). Curvularia lunata rhamnosidase can remove L-rhamnose from a number of steroidal saponins (Feng et al. 2007).

Production of ginsenosides
Glycosyl groups of ginsenosides can be cleaved by α-L-rhamnosidase to produce ginsenosides with improved
activities. The ginsenoside-Rh1 obtained from ginsenosides-Rg2 using α-L-rhamnosidase exhibits anti-cancer activity (Ribeiro 2011).

Antibiotics preparation
The glycopeptides antibiotic chloropolysporin C is prepared from the related compound chloropolysporin B by enzymatic hydrolysis using Rhase (Sankyo 1988). It exhibits antibacterial activity, useful in the treatment and prophylaxis of infections, and is a growth-promoting agent for animals.

Prunin preparation
The flavonoid prunin is produced from naringin using α-L-rhamnosidase activity. It possesses anti-inflammatory and antiviral activity against DNA/RNA viruses (Kaul 1985) (Fig. 1).

Flavonoids deglycosylation
Deglycosylation of flavonoids in Cleome arabica leaf extracts (CALE) with naringinase may be an important therapeutic factor in the treatment of chronic diseases (Bouriche and Arnhold 2010).

CONCLUSIONS
This review highlights the information of α-L-rhamnosidases from yeast origin. In our knowledge concerning rhamnosidases, there is a wide gap between yeast and other microorganisms. At present about yeast strains, only a few reports have addressed the enzymatic activity, fermentative production on a large scale, characterization, and its subsequent application. Also, its structural, functional, and molecular biology aspects have not been reported. Although several rhamnosidase coding genes are identified from various bacteria, none have been used for the commercial production of the enzyme. The studies in the above directions will be required for this purpose. Accordingly, there is a need for a process of producing and purifying α-L-rhamnosidase from more efficient new yeast strains which exhibit improved enzyme activity, with enhanced shelf life, high thermal stability, more efficient in food, chemical, and pharmaceutical industries, and non-toxic for human consumption.

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